



5-ARYL- β,γ BUTENOLIDE, A NEW CLASS OF ANTIBACTERIAL DERIVED FROM THE N-ARYL OXAZOLIDINONE DUP 721

A. DENIS*, T. VILLETTE

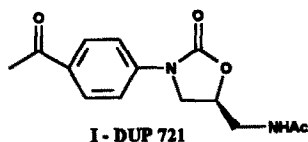
ROUSSEL UCLAF, 102 Route de Noisy, F 93235 Romainville Cedex, France

Abstract :

The synthesis of three heterocyclic analogues of DUP 721 is described. Replacement of the oxazolidinone ring of DUP 721 by a β, γ butenolide gives a compound with similar antibacterial activity.

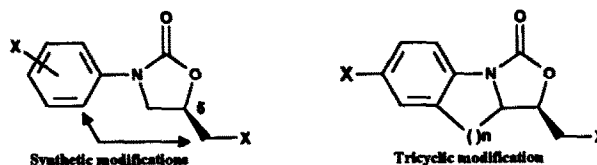
The 3-aryloxazolidinones are a new class of synthetic antibacterial agents discovered by DuPont de Nemours 1a,b,c,d. Different studies concerning the mechanism of action have shown that the oxazolidinones inhibit protein synthesis ². An improvement of the antibacterial and pharmacological profile has been achieved in this family by the synthesis of DUP 721 (figure 1).

Figure 1 :



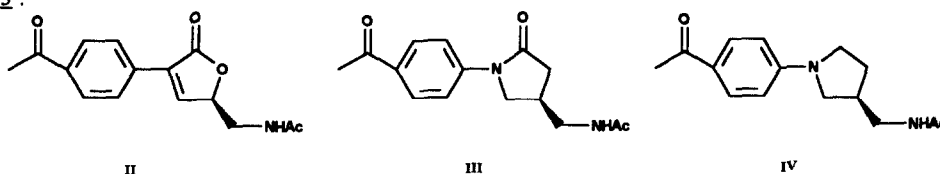
The *in vitro* spectrum of this compound includes activity against susceptible and resistant strains of staphylococci, streptococci, enterococci, anaerobic bacteria and mycobacteria ³. Most of the synthetic modifications have been directed on two positions of the molecule : the aromatic ring ¹ and the 5-methylene (figure 2). In a recent work, some interesting tricyclic modifications were also described (figure 2) ⁴ :

Figure 2 :



Surprisingly, the nature of the central heterocycle has never been explored. We speculated that the very flat oxazolidinone ring could be substituted by a β,γ butenolide ring, or by more flexible compound such as pyrrolidinone or pyrrolidine (figure 3).

Figure 3 :

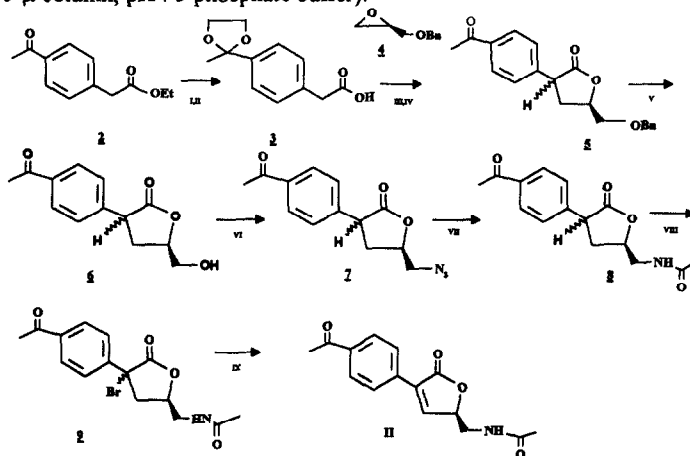


We therefore decided to investigate the influence of these structural modifications on the antibacterial activity.

Synthesis :

Since chiral starting material was available for the synthesis of the lactone II, we chose to synthesize the chiral parent compound of DUP 721 (scheme 1). The synthesis started with the known p.acetylphenylethylacetate **2** ⁵ which was protected by a dioxolane ring and hydrolyzed to yield acid **3**. The lithiated dianion of **3** was reacted with the commercial (S) benzyloxymethyloxirane **4** in THF at 0°C. The resulting γ hydroxy-acid was directly cyclized with simultaneous cleavage of the dioxolane ring by treatment of the crude product with p. toluene sulfonic acid in toluene in the presence of 4 Å molecular sieves. Lactone **5** was obtained in 65 % yield as a 1/1 diastereoisomeric mixture. The benzyl group was removed by hydrogen transfer reaction ⁶ to avoid reduction of the acetyl group (82 % yield). The N-acetyl function was then introduced using standard procedures : mesylation, nucleophilic substitution by NaN₃ and reduction with P(OMe)₃ ⁷ in 52 % overall yield. Bromination of **8** by NBS in CCl₄ gave only 20 % of the desired bromo compound **9** along with unchanged starting material which could be recycled.

The double bond was introduced by elimination of the bromine by treatment of **9** with 3.3 eq. of dry pyridine in toluene at 80°C. The (R) lactone II was obtained in 65 % yield, with a 80 % optical purity (determined by chiral HPLC : HSA 5 μ column, pH : 5 phosphate buffer).

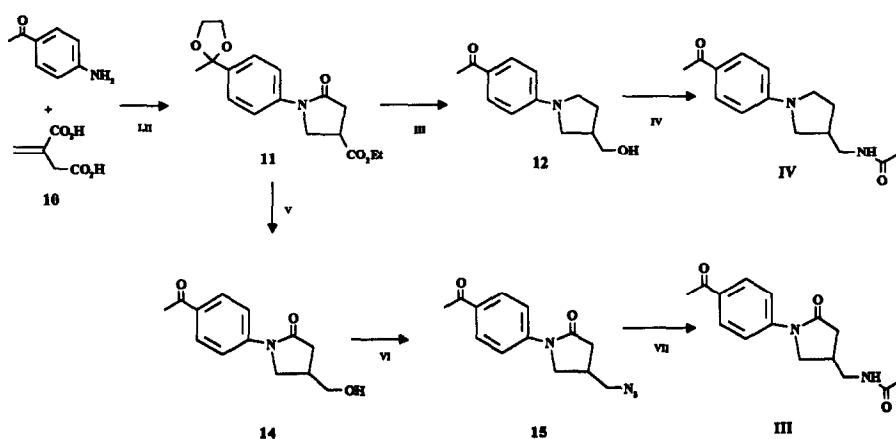


Scheme 1 : β,γ Butenolide synthesis

(I) HO(CH₂)₂OH, HC(OEt)₃, APTS cat., 50°C, 93%; (II) -a) NaOH 2N, EtOH, 50°C - b) AcOH, 5°C, 95%; (III) -a) LDA, THF, -5°C - b) **4**, 0°C to 24°C; (IV) APTS cat., toluene, 4 Å molecular sieves, 65%; (V) Pd(OH)₂, cyclohexene, EtOH, Rfx., 82%; (VI) -a) CH₃SO₂Cl, Et₃N, CH₂Cl₂ - b) NaN₃, DMF, 80°C, 75%; (VII) -a) P(OMe)₃, MeOCH₂OMe, Rfx. - b) HCl 6N, Rfx. - c) NaOH 2N, Ac₂O, THF/H₂O, 69%; (VIII) NBS, Ph(COO)₂, CCl₄, Rfx., 20%; (IX) Pyridine, Toluene, 80°C, 65%.

The heterocycles III and IV were synthesized as racemates from itaconic acid **10** using described procedures ⁸ (scheme 2).

Fusion of p.acetylaniline with itaconic acid **10** followed by protection of the carbonyl, and esterification gave the carboxypyrrolidone **11** in 48 % yield. Reduction with NaBH_4 or LiAlH_4 provided pyrrolidinone **14** (96 % yield) and pyrrolidine **12** (97 % yield) respectively. Procedures analogous to the ones used for the lactone II led to the desired N-acetyl pyrrolidinone III in 58 % yield. Surprisingly, the azidomethyl pyrrolidine was directly reduced probably by thermal decomposition of the azido group during work up of the reaction. However, acetylation of the crude product gave the expected N-acetyl pyrrolidine IV in 25 % yield.



Scheme 2 : synthesis of pyrrolidones III and pyrrolidine IV

(I) -a) 140°C , neat -b) EtOH , CH_3COCl , Rfx, 48% ; (II) $\text{HO}(\text{CH}_2)_2\text{OH}$, $\text{HC}(\text{OEt})_3$, APTS cat., 65°C , 100% ;
 (III) -a) LiAlH_4 , THF, Rfx. -b) HCl 6N, 97% ; (IV) -a) $\text{CH}_3\text{SO}_2\text{Cl}$, Et_3N , CH_2Cl_2 -b) NaN_3 , DMF, 85°C
 -c) Ac_2O , THF/ H_2O , 24% ; (V) -a) NaBH_4 , MeOH -b) HCl 6N, 76% ; (VI) -a) $\text{CH}_3\text{SO}_2\text{Cl}$, Et_3N , CH_2Cl_2
 -b) NaN_3 , DMF, 85°C , 81% ; (VII) MeOCH_2OMe , $\text{P}(\text{OMe})_3$, Rfx. -d) HCl 6N, Rfx.
 -e) NaOH 2N, Ac_2O , THF/ H_2O , 72% .

Biological results :

The three compounds have been tested on both susceptible and resistant staphylococci and enterococci ; DUP 721 was used as a reference compound ⁹ (table 1).

Lactone II was as active as DUP 721 and exhibited a similar antimicrobial spectrum : active against Gram (+) strains (susceptible and resistant), MIC \leq 2.5 - 5, inactive against Gram (-) strains, MIC $>$ 40. III and IV were inactive.

Table 1 : In vitro antibacterial activity

MIC (μ g/ml)	DUP 721	II	III	IV
<i>Staphylococcus aureus</i> SG 511 ^b	1.2	1.2	$>$ 40	$>$ 40
<i>Staphylococcus aureus</i> O11CB20 ^c	1.2	1.2	$>$ 40	$>$ 40
<i>Staphylococcus epidermidis</i> 012GO39 ^c	2.5	2.5	$>$ 40	$>$ 40
<i>Streptococcus pyogenes</i> 77A ^b	0.6	2.5	$>$ 40	$>$ 40
<i>Streptococcus faecium</i> M78L ^b	2.5	5	$>$ 40	$>$ 40
<i>Streptococcus faecium</i> 02D3IP1 ^d	$<$ 1.2	2.5	$>$ 40	$>$ 40
<i>Streptococcus faecalis</i> 02D2DU17 ^d	2.5	2.5	$>$ 40	$>$ 40
<i>Escherichia coli</i> UC 1894	$>$ 40	$>$ 40	$>$ 40	$>$ 40

a Minimum Inhibitory Concentrations (MICs) were determined by a standard agar dilution method.

b Sensitive strains

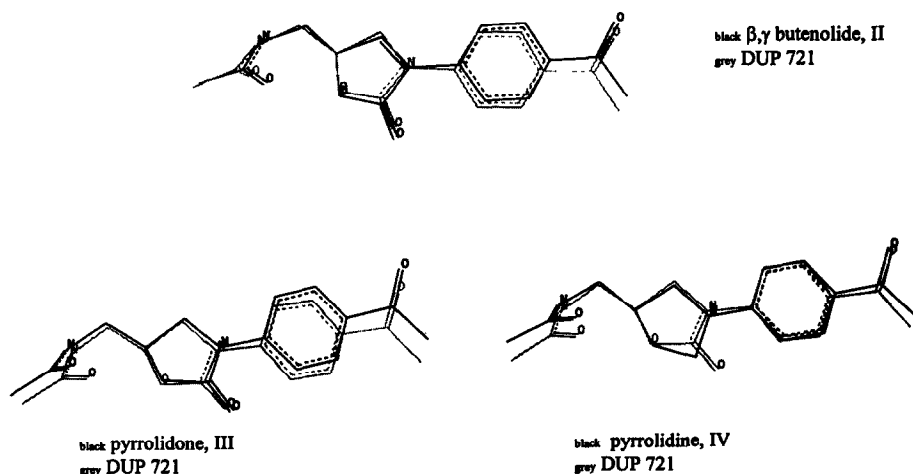
c Oxacilline Resistant

d Vancomycin Resistant

A molecular modelling analysis (Discover ® Software) showed very good correlation between the three compounds and DUP 721 (figure 4). This confirmed the hypothesis of a requirement for coplanarity of the aromatic and the heterocyclic rings in the case of the active β , γ butenolide II.

The lack of activity of the aminoheterocycles could be better explained in terms of functional changes. Indeed, the more basic pyrrolidine IV is devoid of carbonyl and the amide function of the pyrrolidinone III is more polar than its carbamate or lactone counterpart in I or II.

These two examples illustrated the importance of the presence and the nature of the carbonyl group for the activity.

Figure 4 : Superposition of II, III, IV and DUP 721**Conclusion :**

This study demonstrates for the first time that the basic skeleton of the 3-aryl oxazolidinones antibiotics could be extended to a new structure : a 5-aryl- β,γ butenolide.

This lactone exhibits a similar antibiotic activity and offers an opportunity to further modifications in this protein synthesis inhibitor family.

Acknowledgment :

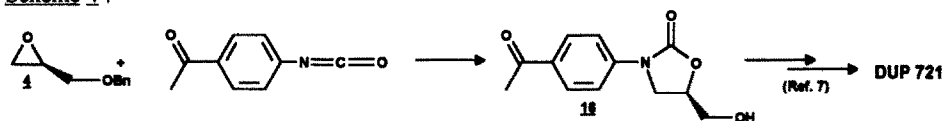
P. Brotto for Molecular Modelling.

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- 9) (S) DUP 721 was synthesized following a modified procedure of Wang et al. ⁷ (scheme 4) :

Scheme 4 :



The (S) glycidyl butyrate was replaced by the (S) benzyloxymethyloxirane 4, which was reacted with p. acetylisocyanate in xylene with LiBr, $n\text{Bu}_3\text{PO}$ at reflux in 98 % yield ; the benzyl group was then removed by treatment with $\text{Pd}(\text{OH})_2$, cyclohexene in refluxing ethanol in 83 % yield to afford the described 5-hydroxymethylene oxazolidinone intermediate 16.

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